

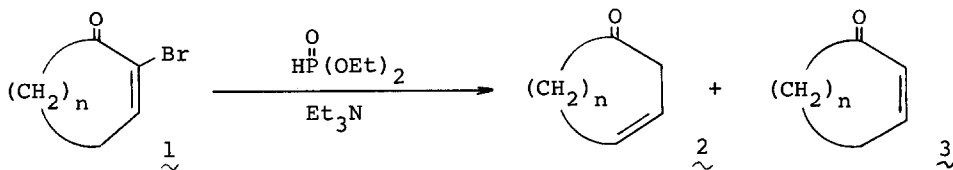
NOVEL REDUCTION OF  $\alpha$ -BROMO- $\alpha,\beta$ -UNSATURATED KETONES.  
SELECTIVE SYNTHESIS OF  $\beta,\gamma$ -UNSATURATED KETONES

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Summary: Treatment of  $\alpha$ -bromo- $\alpha,\beta$ -unsaturated ketones or 1,1-dibromo-2-trimethylsilyloxycyclopropanes with diethyl phosphite and triethylamine affords  $\beta,\gamma$ -unsaturated ketones selectively.

In our preceding paper,<sup>1</sup> diethyl phosphite-triethylamine is proven to be a versatile reagent for reduction of gem-dibromocyclopropanes and gem-dibromoalkenes to the corresponding monobromo derivatives, respectively. We describe herein that the reaction of  $\alpha$ -bromo- $\alpha,\beta$ -unsaturated ketones with diethyl phosphite and triethylamine gives  $\beta,\gamma$ -unsaturated ketones selectively. This process comprises reduction of the bromide and deconjugation of an olefinic moiety.



Typical procedure is as follows. To a solution of 0.945 g (5.0 mmol) of 2-bromo-2-cycloheptenone and 0.829 g (6.0 mmol) of diethyl phosphite was added 0.607 g (6.0 mmol) of triethylamine at room temperature, and the mixture was stirred for 15 h. Ether (50 mL) was added and filtered to remove  $Et_3N \cdot HBr$ . After concentration of the filtrate, the residue was chromatographed on a silica gel column eluting with benzene to produce 0.350 g of 3-cycloheptenone (2a) in 64% yield. A small amount of 2-cycloheptenone (3a) was separated as a by-product. Some results are summarized in Table 1.

The present reaction was applicable to acyclic  $\alpha$ -bromo- $\alpha,\beta$ -unsaturated ketones. For example, 5-bromo-4-decen-6-one<sup>2</sup> was treated with diethyl phosphite and triethylamine at room temperature for 16 h to yield (E)-3-decen-6-one (61%), being contaminated with (E)-4-decen-6-one (3%). Another important characteristic is stereoselectivity of the olefin geometry of  $\beta,\gamma$ -unsaturated ketones; in the reaction of 2-bromo-2-cycloundecenone<sup>2</sup> or 5-bromo-4-decen-6-one, the corresponding (E)- $\beta,\gamma$ -unsaturated ketone was produced.

Table 1. Preparation of  $\beta,\gamma$ -Unsaturated Ketones

$\tilde{1}$	HP(O)(OEt) <sub>2</sub> (equiv.)	Et <sub>3</sub> N (equiv.)	Reaction Temp.	Time	Products % $\tilde{2} + \tilde{3}$
2-bromo-2-cycloheptenone	1.2	1.2	rt	15 h	66 (96:4)
2-bromo-5-methyl-2-cycloheptenone	1.2	1.2	rt	20	41 (91:9)
2-bromo-2-cyclooctenone	3	1.5	80°C	3.5	47 (81:19)
2-bromo-2-cycloundecenone <sup>a</sup>	3	2	80°C	1	84 <sup>b</sup> (90:10)

a. The olefin geometry was not determined. b. An E isomer of  $\tilde{2}$  or  $\tilde{3}$  was produced.

$\alpha$ -Bromo- $\alpha,\beta$ -unsaturated ketones were given by treatment of 1,1-dibromo-2-siloxycyclopropanes with diethyl phosphite. The reaction of 7,7-dibromo-1-trimethylsiloxybicyclo[4.1.0]heptane (4a) with diethyl phosphite at 50°C for 2 h produced 2-bromo-2-cycloheptenone ( $\tilde{1a}$ ) in 90% yield. The same elimination in pyrolysis or in the presence of acid was reported.<sup>3</sup> This finding made it possible to prepare  $\beta,\gamma$ -unsaturated ketones from 1,1-dibromo-2-siloxycyclopropanes in one step. Treatment of 4a with 4 equiv. of diethyl phosphite and 3 equiv. of triethylamine at room temperature for 5 days gave  $\tilde{2a}$  and  $\tilde{3a}$  in 37% yield ( $\tilde{2a}:\tilde{3a}=78:22$ ), though the selectivity and yield of  $\tilde{2a}$  were low.

The mechanism of the present conversion has not been clarified. Isomerization between  $\tilde{2a}$  and  $\tilde{3a}$  was not observed in contact with diethyl phosphite and triethylamine at room temperature. As one of the plausible reaction paths, the  $\alpha$ -bromo- $\beta,\gamma$ -unsaturated ketone, arising from the intervention of the dienol, is subjected to debromination with diethyl phosphite and triethylamine to yield the  $\beta,\gamma$ -unsaturated ketone  $\tilde{2}$ .

$\alpha$ -Bromo- $\alpha,\beta$ -unsaturated ketones are alternatively prepared from gem-dibromocyclopropanes.<sup>4</sup> The present process provides a convenient method for the preparation of  $\beta,\gamma$ -unsaturated ketones. A full scope of this reaction is being investigated.

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#### References and Note

1. T. Hirao, T. Masunaga, Y. Ohshiro, and T. Agawa, *J. Org. Chem.*, **46**, 3745 (1981).
2. The olefin geometry was not determined.
3. P. Amice, L. Blanco, and J. M. Conia, *Synthesis*, **1976**, 196.
4. M. Balci and W. M. Jones, *J. Am. Chem. Soc.*, **102**, 7607 (1980).

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